



Advanced Nanoporous Composite Materials for Industrial Heating Applications

Towards Low-Cost Nanostructured Refractories

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IHEA Forum - ORNL

Industrial Needs for Insulation

- Industrial hearth, tank, and pot furnaces used by various IoF's (e.g. Glass) can benefit from improved insulations in critical areas
 - Regenerators, ports, crowns, forebay channel, etc.
- Current insulations used: firebricks, fiber bats, composite materials, microporous insulation
- Desired performance improvements:
 - Lower thermal conductivity
 - Space savings
 - Longer life times
 - Increased corrosion resistance
 - Lower cost

Aerogels: Background & Process

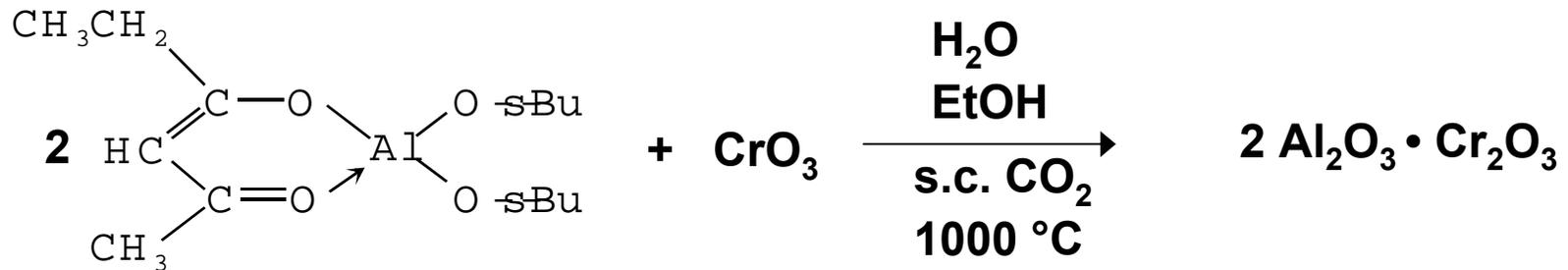
- Aerogels: Nanoporous, open-celled solids formed by controlled removal of the liquid phase from a gel.
 - First prepared by Samuel Kistler in 1931
- Typical preparation:
 - Sol-Gel formation of wet gel
 - Hydrolysis-condensation of Alkoxides
 - Organic polymerization
 - Other colloid-forming methods
 - Supercritical drying
 - Alcohol drying
 - CO₂ substitution-drying

Challenges for High Temperature Uses

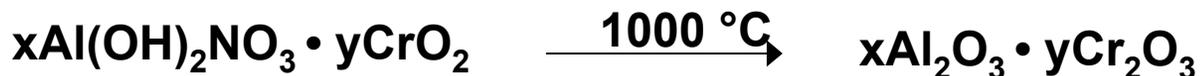
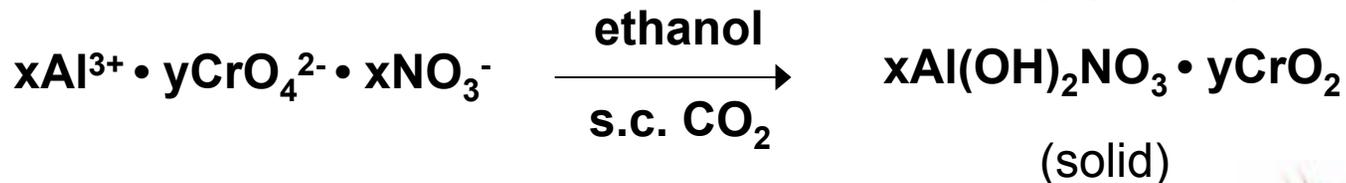
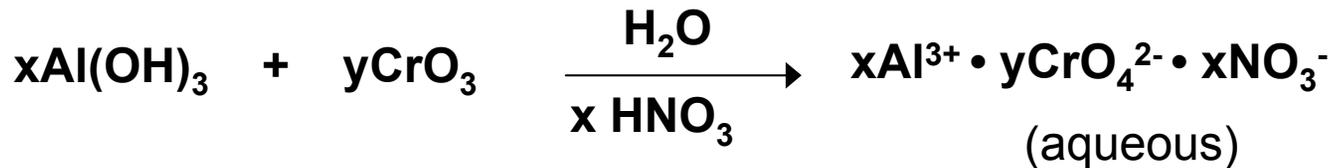
- Must withstand temperatures of 700-1500 °C
 - Increase sintering resistance
- Must be chemically inert
- Must show reduced thermal conductivity
 - Target: 0.01-0.10 W/m-K
 - Block solid, gaseous, and *Radiative* heat transfer modes
- $x(\text{Al}_2\text{O}_3) \cdot y(\text{Cr}_2\text{O}_3)$ or $x(0.94\text{Al}_2\text{O}_3/0.06\text{SiO}_2) \cdot y(\text{Cr}_2\text{O}_3)$
- Must be affordable
 - Target: \$5-20 per board foot, Composition:

Two Routes to Al₂O₃-Cr₂O₃ Aerogels

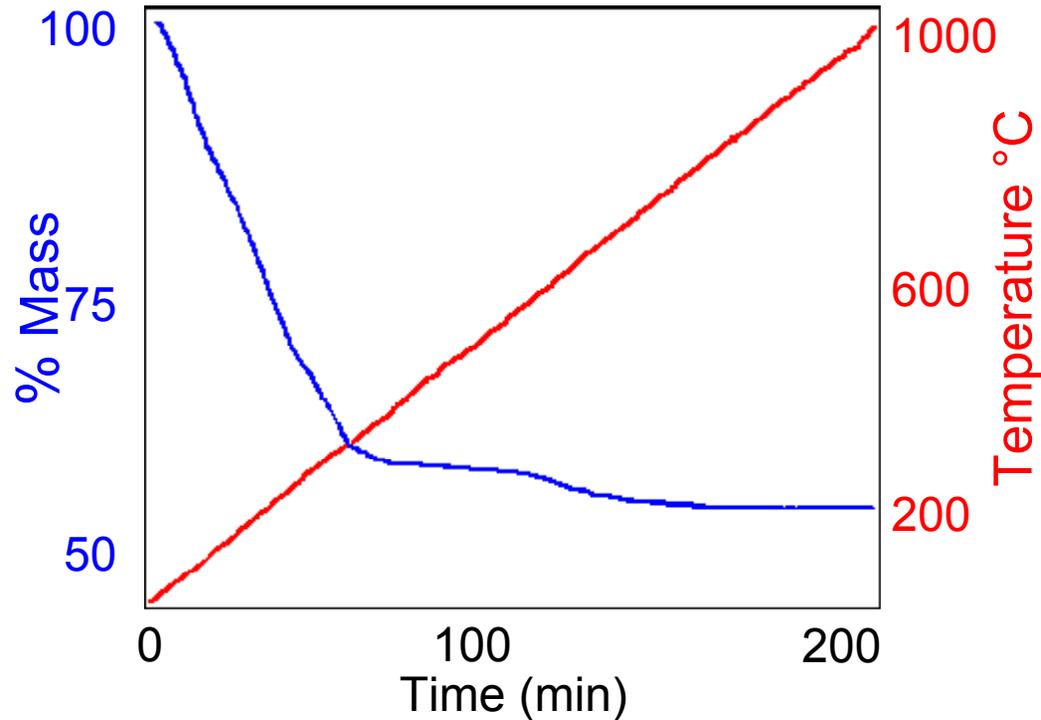
- Aluminum alkoxide/CrO₃



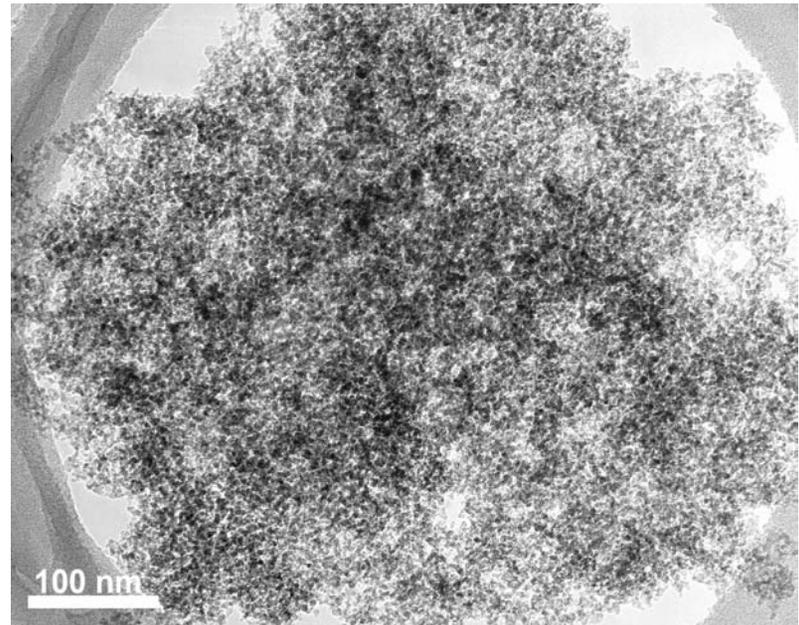
- Acid-Base



Microstructure After Firing



TGA (air) plot of
alkoxide-derived aerogel



TEM image of alkoxide-derived
aerogel fired at 1000 °C

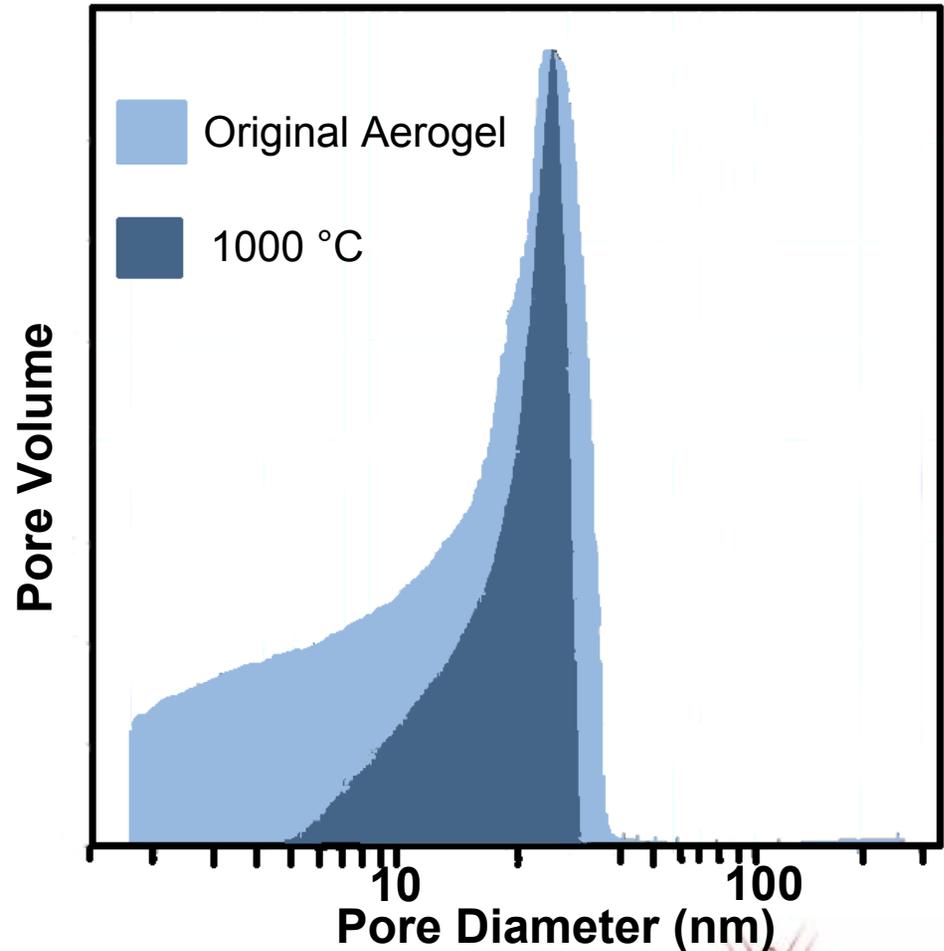
Effect of Temperature on Surface Area

- Surface areas (BET) of various compositions (acid-base route) before and after firing (m^2/g)

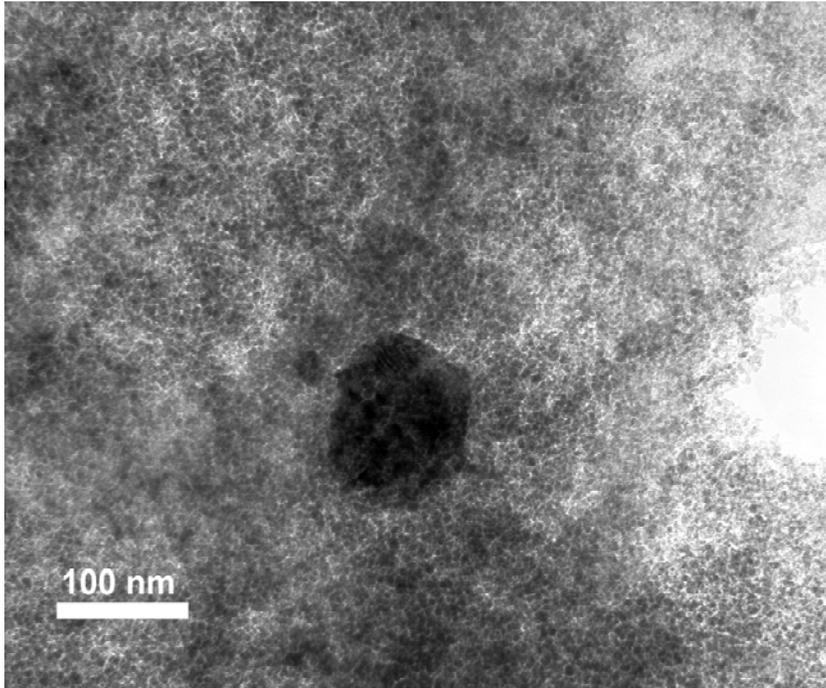
Compound	Neat aerogel	450 °C	1000 °C
Cr_2O_3	290	13	13
$\text{Al}_2\text{O}_3 \cdot 2\text{Cr}_2\text{O}_3$	270	180	41
$\text{Al}_2\text{O}_3 \cdot \text{Cr}_2\text{O}_3$	260	160	44
$2\text{Al}_2\text{O}_3 \cdot \text{Cr}_2\text{O}_3$	240	170	64
$2(0.94\text{Al}_2\text{O}_3 \cdot 0.06\text{SiO}_2) \cdot \text{Cr}_2\text{O}_3$	350	—	130

Porosity After Thermal Processing

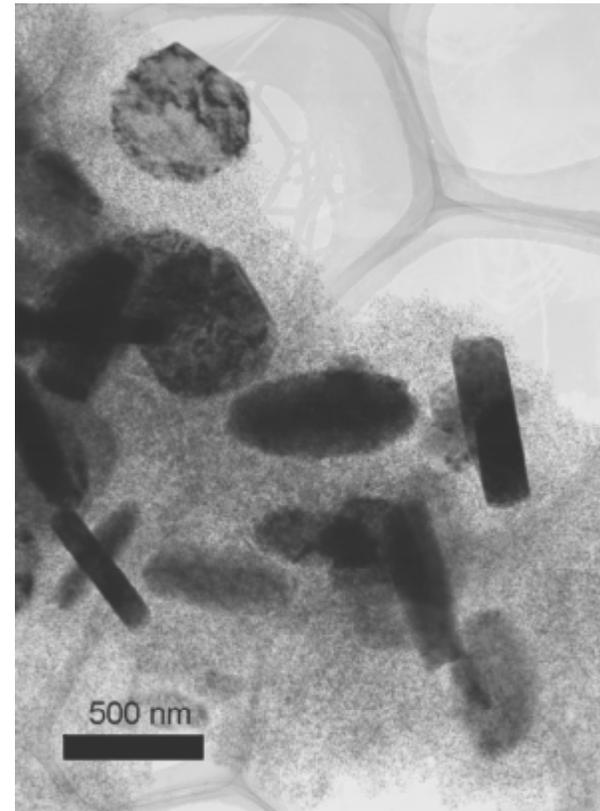
- Peak of Pore size distribution is ~ 26 nm
- Considerable pore volume between 2-10 nm
- Thermal treatment closes smaller pores
- Peak remains at ~ 26 nm



TEM Images of $\text{Al}_2\text{O}_3\text{-Cr}_2\text{O}_3$ Aerogels

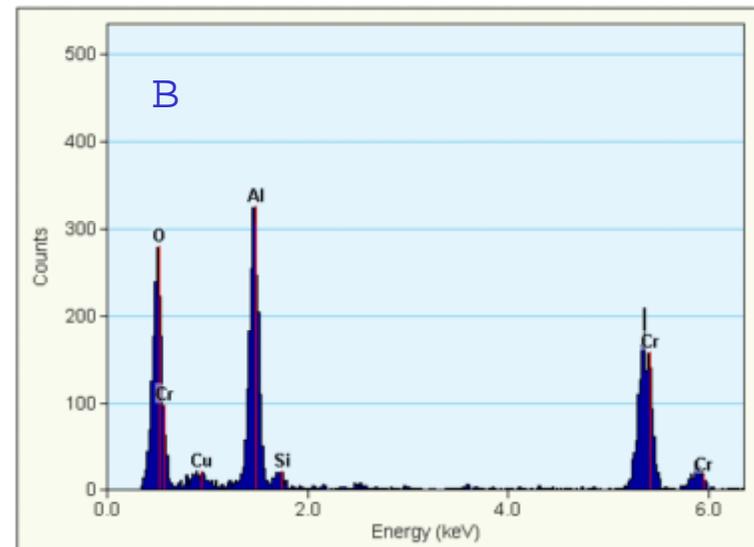
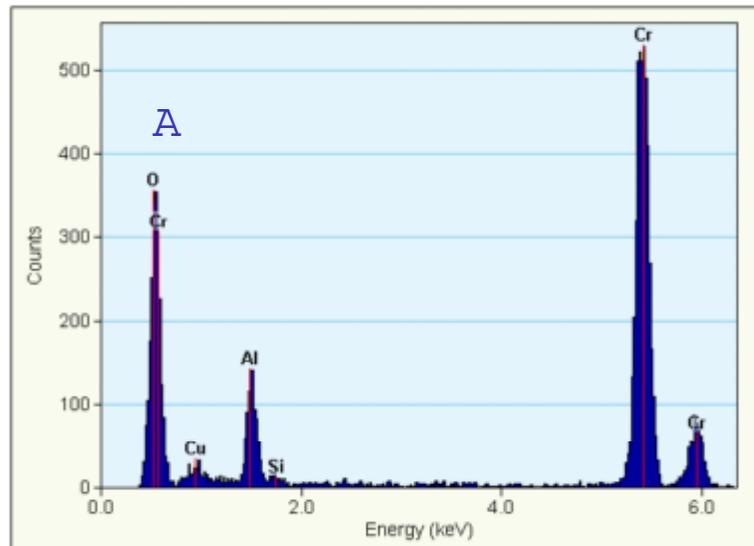
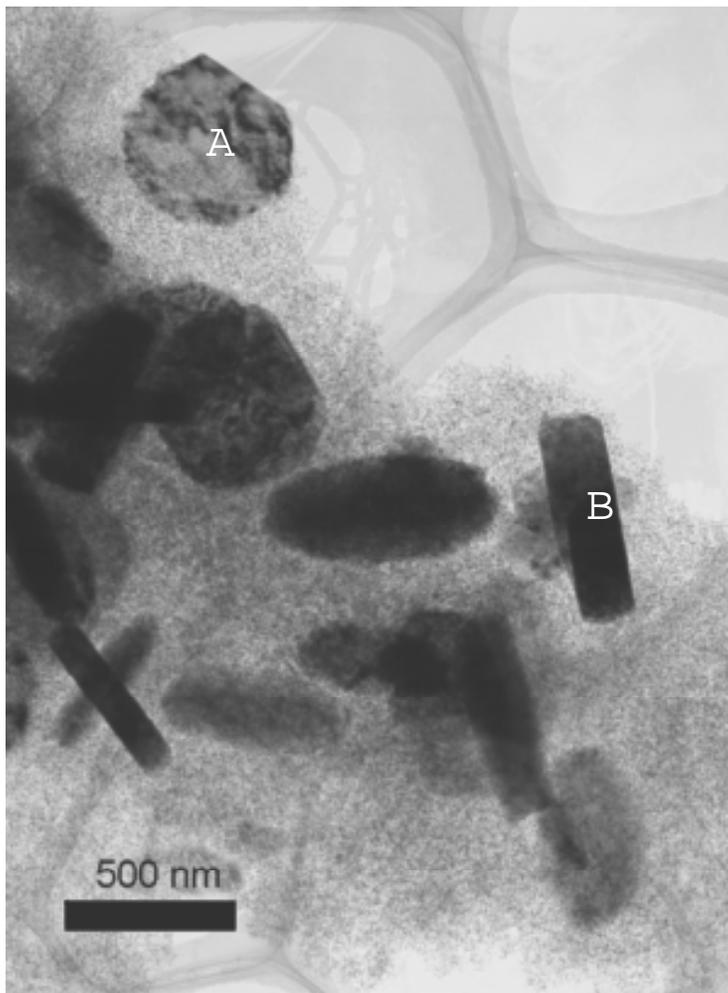


**Alkoxide
route**



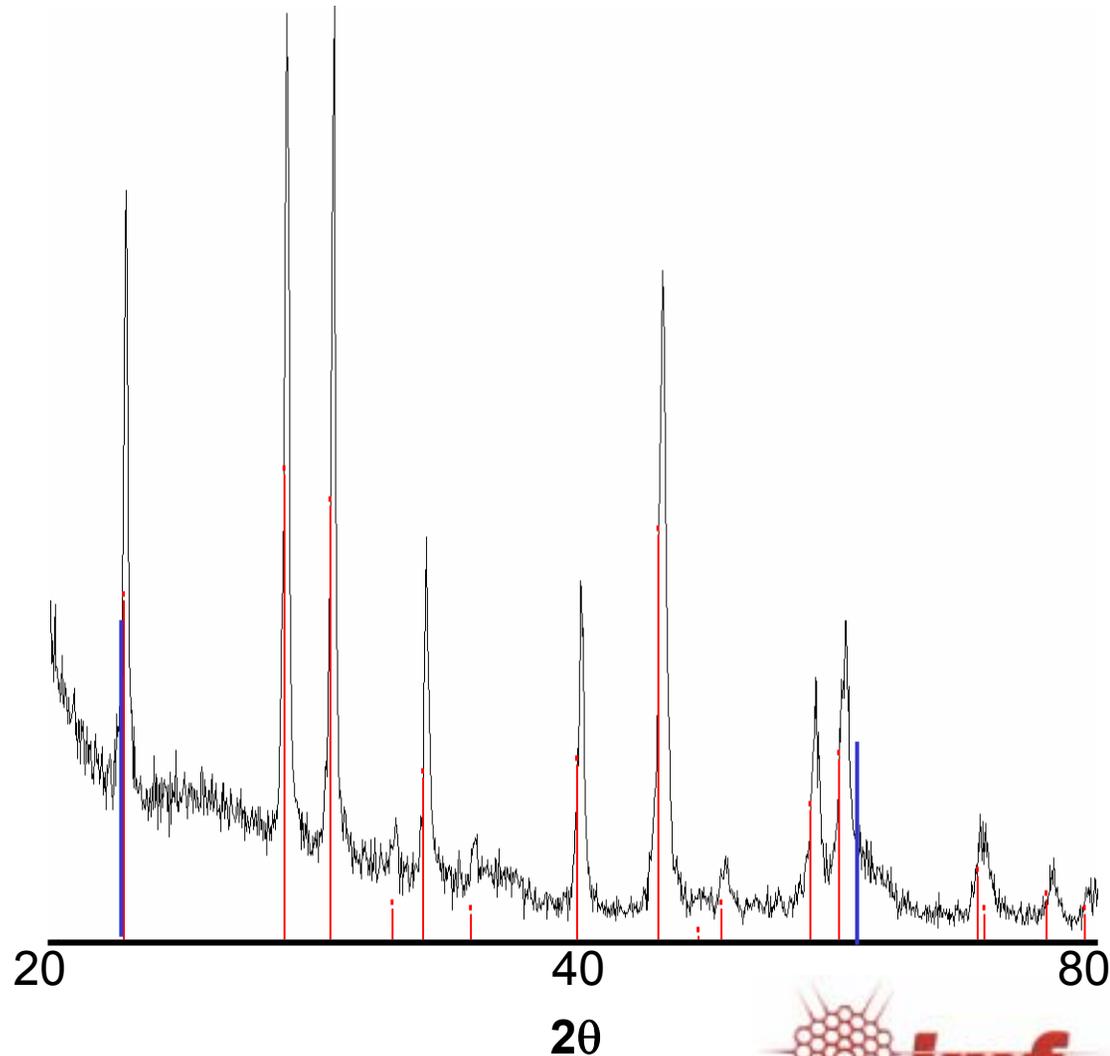
Acid- Base route

EDX Indicates Two Phases



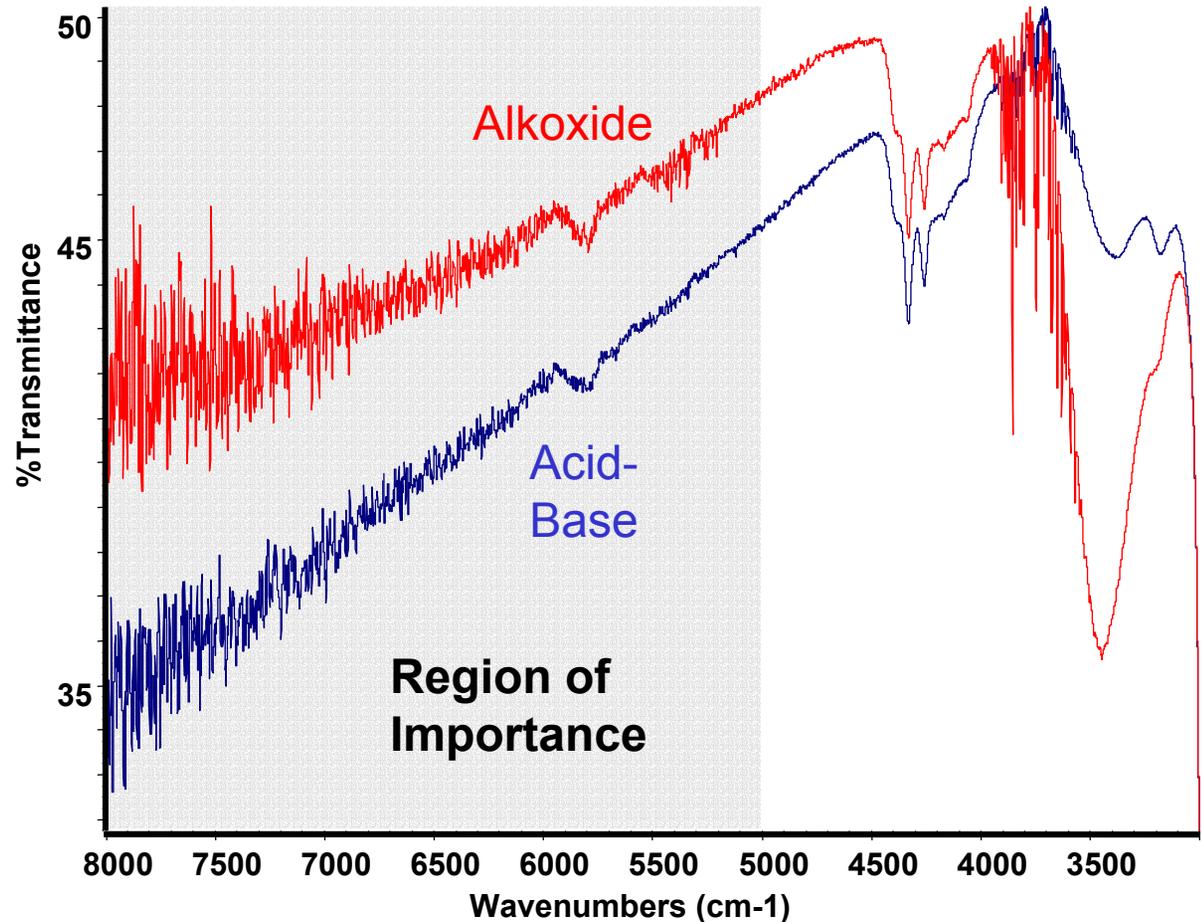
X-Ray Identification

- Acid-base derived aerogel shows one primary crystalline phase:
 - Eskolaite, Cr_2O_3
- Second phase may also be present:
 - $\text{Al}_{1.4}\text{Si}_{0.3}\text{O}_{2.7}$



Opacity of Al-Cr Aerogels

- 0.1-1.5 micron crystal inclusions scatter incident IR radiation
- Acid base route gives higher # density of crystals
- Shows lower transmittance in area of interest
- Material is largely self-opacified



Raw Materials Costs

For 1 board-foot @ 0.35 g/cm^3 (~8% solids); bulk or semi-bulk

Worst case:

Item	Cost/board-foot (\$)
Aluminum oxide	58
CrO_3	1.36
Tetraethylorthosilicate	0.54
ZrO_2 fiberopacifant (1% v/v)	58

Total: \$118.00

Best case:

Item	Cost/board-foot (\$)
Aluminum trihydroxide	0.23
CrO_3	1.36
Tetraethylorthosilicate	0.54

Total: \$2.13



Future Work

- Continue compositional study and characterization of alumina-chromia aerogels to achieve lower density and better opacification
- Evaluate best approach for using this material in a commercial insulation product
 - Compacted Monoliths,
 - Additions to Composite products
- Cost estimates for physical processing
- Prototyping for glass furnace application
- Expand program to look at various other routes to nanostructured materials derived from commodity raw materials

